

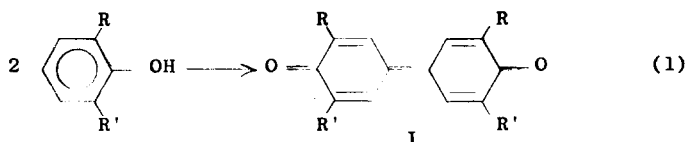
DEHYDROGENATION REACTIONS WITH DIPHENOQUINONES

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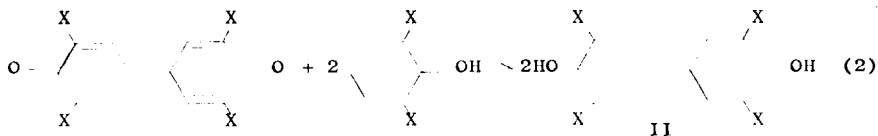
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The oxidative coupling with oxygen, of 2,6-disubstituted phenols in which at least one of the substituents is a large, bulky group (such as t-butyl) yields diphenoquinones (I) in high yield (1,2). High molecular weight polyethers can be prepared by oxidative coupling of 2,6-disubstituted phenols



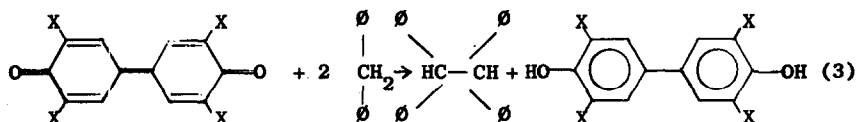
if the substituents are small (2), however, under the proper conditions the diphenoquinone is the major product (3). These diphenoquinones are readily reduced to the hydroquinones (4). We have now found that the hydroquinones are produced in high yield by an oxidation-reduction reaction involving a phenol and a diphenoquinone (eqn. 2). The reaction is catalyzed by amines,



acids and aluminum salts. For example, to a Carius tube was

added 1.29 g. (0.0032 moles) of 3,3',5,5'-tetra-t-butyl diphen-
 quinone, 1.5 g. (0.0073 moles) of 2,6-di-t-butyl phenol and
 1 g. of pyridine. The tube was flushed with nitrogen, sealed
 and placed in a furnace at 250°C. In ten minutes the reaction
 mixture was pale yellow. The reaction mixture was cooled and
 diluted with ether and washed with dilute hydrochloric acid to
 remove pyridine. The ether was evaporated and 10 ml. of hot
 ethanol was added. On cooling there deposited 2.51 g. (0.0061
 moles; 95% yield) of pale yellow crystals, m.p. 186°C. (lit.
 m.p. 184-185° (4)). This material is identical with an authentic
 sample of 2,2',6,6'-tetra-t-butyl p,p'-biphenol (II; R=t-butyl).
 When a more strongly basic amine (e.g., trimethylamine) is used
 the reaction proceeds readily at 150°C. and reaction of the more
 powerful oxidant 3,3',5,5'-tetramethyldiphenoquinone with 2,6-
 dimethylphenol proceeds rapidly at 100°C. to yield principally
 (80%) the biphenol (I; R=R'=CH₃).

The diphenoquinones also react with hydrocarbons such as
 diphenylmethane as shown in equation 3.



To a 100 ml. flask equipped with an air condenser was
 added 40 ml. of diphenylmethane and 5 g. (0.012 moles) of
 3,3',5,5'-tetra-t-butyl diphenoquinone. The solution was
 refluxed for one hour (260°C.) then cooled and 50 ml. of ethanol
 was added. The solid that deposited was recrystallized from
 ethanol-chloroform to yield 2.6 g. (0.0078 moles; 65% yield)

of 1,1,2,2-tetraphenylethane, m.p. 208° (lit. 210° (5)).

Further work on the use of these diphenoquinones in dehydrogenation reactions is in progress.

References

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